476. Dimethinmerocyanines derived from 2-Substituted Azol-5-ones.*

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A number of S-esters (V), including long-chain alkyl esters, of N-dithiocarboxyglycine have been prepared, cyclised by the method of Cook, Harris, Heilbron, and Shaw (J., 1948, 1056), and converted into a series of dimethin*mero*cyanines (cf. Cook, Harris, and Shaw, J., 1949, 1435) containing the 2-alkylthiothiazol-5-one nucleus. Analogous O-esters (XIV) of N-thioncarboxyglycine were readily obtained by the condensation of potassium aminoacetate with ethyl alkoxydithioformates. Cyclisation of these with acetic anhydride gave a series of 2-alkoxythiazol-5-ones (XV), or, in the presence of ethyl orthoformate, 2-alkoxy-4-ethoxymethylenethiazol-5-ones (XVI) both of which were converted into a series of dimethin*mero*cyanines. Similarly N-acylthiohydantoic acids were cyclised by phosphorus tribromide to give what are believed to be 2-acylamidothiazol-5-one hydrobromides which also contain a reactive methylene group giving rise to a series of dimethin*mero*cyanines.

The condensation of N-dithiocarbethoxyglycine with secondary amines led to NN-disubstituted N'-carboxythioureas (XIX) which were cyclised by phosphorus tribromide to 2-di-alkyl(or -aryl)aminothiazol-5-ones which also readily formed dimethin*merocyanines*.

The series of dimethin*merocyanines* derived from 2-phenyloxazol-5-one (Cook *et al., loc. cit.,* 1949) was extended.

BECAUSE of their usefulness as photographic optical sensitisers the class of dyes known as dimethinmerocyanines (I) has been widely investigated. In (I), R is usually alkyl, R' is hydrogen, alkyl, or aryl, n is 0 or 1, and A and B represent the atoms necessary to complete a 5- or 6-membered ring or rings. In the majority of known dyes the ketonic nucleus carries a substituted amino-group adjacent to the ketonic group (*i.e.*, X = NR). Keyes and Brooker (B.P. 518,904), however, obtained a number of dimethinmerocyanines containing the 2-phenyl-oxazol-5-one nucleus (II; X = O, Y = Ph), and recently Cook *et al.* (*J.*, 1949, 1435) obtained similar dyes by a different method and also related dyes from the 2-substituted thiazol-5-one



nucleus (II; X = S, Y = Ph, $\cdot S \cdot CH_2Ph$, $\cdot S \cdot CO_2Et$) The latter authors prepared their dyes by the known procedure of condensing the 4-ethoxymethylene derivative (III; R' = H) of (II) with the reactive methyl derivative of the required cyclic quaternary ammonium salt in the presence of alcoholic triethylamine.

In view of the possible photographic application of such dyes a larger number have been synthesised (see Experimental) containing the 2-phenyloxazol-5-one, 2-ethylthiothiazol-5-one, 2-benzylthiothiazol-5-one, and 2-mercaptothiazol-5-one nuclei (cf. II; X = S, Y = SH) (Cook, Heilbron, and Levy, J., 1948, 201). These dyes were obtained both by Cook's method

* Patent applications pending.

and by the more usual method of condensing (II) with the 2-acetanilidovinyl derivative of the cyclic quaternary ammonium salt in alcoholic triethylamine. An extension of the series of dyes was made possible (see Experimental) by the synthesis of a number of further S-esters (V) of N-dithiocarboxyglycine. Thus, using Körner's method (*Ber.*, 1908, **41**, 1091), the p- and *m*-nitrobenzyl esters (V; $\mathbf{R''} = \cdot \mathbf{CH}_2 \cdot \mathbf{C}_6 \mathbf{H}_4 \cdot \mathbf{NO}_2$) were obtained and cyclised by phosphorus tribromide to the required hydrobromides of 2-p- and 2-*m*-nitrobenzylthiothiazol-5-one (VI). It was also possible to make long-chain alkyl esters (V), by refluxing the required alkyl bromide with the potassium salt of N-dithiocarboxyglycine. In this way the *n*-hexyl, *n*-heptyl, *n*-octyl, *n*-decyl, *n*-dodecyl, and 3-phenylpropyl esters (V) were readily obtained. The salts of the longer-chain esters possess some surface activity. All of these esters were readily cyclised by

$$(V.) \qquad R''S_2C\cdot NH\cdot CH_2\cdot CO_2H \longrightarrow \begin{array}{c} S \longrightarrow CO \\ | & | \\ R''S\cdot C_N \swarrow CH_2 \end{array} (VI.)$$

acetic anhydride or phosphorus tribromide to the required thiazol-5-one (VI) or its hydrobromide (VI; $R'' = C_{12}H_{25}$) was characterised by conversion of its hydrobromide into 2-*n*-dodecylthio-5-hydroxy-4-phenylazothiazole (IV; $R = S \cdot C_{12}H_{25}$) by gently warming it with diazoaminobenzene in ethanol (see also Knott and Williams, *J.*, 1951, 1586, for similar reactions of diazoaminobenzene with reactive methyl groups).

All of the above thiazol-5-ones or their 4-ethoxymethylene derivatives were readily converted into dimethin*mero*cyanine dyes. It has also been found that chain-substituted dyes (I; R' =Me or Et) are readily obtained from the intermediates (III; R' = Me or Et) prepared by cyclisation of the necessary glycine with acetic anhydride in the presence of ethyl orthoacetate or ethyl orthopropionate. Although these intermediates were not isolated in the thiazol-5-one series, crystalline 4-1'-ethoxyethylidene-2-phenyloxazol-5-one (III; R' = Me, X = O, Y = Ph) * (cf. "The Chemistry of Penicillin," Princeton Univ. Press, 1949, p. 803, for III, R' = H) was isolated in good yield.

In view of the strong sensitising properties of many of the foregoing dyes we prepared analogues with other substituents in the 2-position of the thiazol-5-one ring, notably 2-acylamido-, 2-alkoxy-, and 2-*tert*.-amino-groups. The requisite thiazolones were hitherto unknown.

2-Acylamidothiazol-5-ones (X).—Wheeler, Nicolet, and Johnson (Amer. Chem. J., 1911, 46, 456) obtained N-acetylthiohydantoic acid (N-N'-acetylthiocarbamylglycine) (VIII; R = Me) by heating potassium aminoacetate with ethyl acetamidodithioformate (VII; R = Me). They obtained the benzamido-analogue (VIII; R = Ph) by a similar process. On heating these



acids with a mixture of acetic anhydride, acetic acid, sodium acetate, and benzaldehyde they obtained a compound which they formulated as 1-acetyl-5-benzylidenetetrahydro-4-keto-2-thioglyoxaline (the benzylidene derivative of IX; R = Me) since on hydrolysis with cold 10% potassium hydroxide it yielded 5-benzylidenetetrahydro-4-keto-2-thioglyoxaline which was synthesised unambiguously from 2-thiohydantoin and benzaldehyde.

It will be observed that (VIII) is an acylamide of N-dithiocarboxyglycine, the S-ester of which was used by Cook in the synthesis of 2-alkylthiothiazol-5-ones, and that (VIII) is thus the intermediate required for cyclisation to 2-acylamidothiazol-5-ones. The above work (Wheeler *et al.*) indicates, however, that ring closure occurs between the carboxy-group and the amido-nitrogen atom and not between the former and the sulphur atom. Treatment of (VIII; R = Me) with acetic anhydride under conditions used by Cook for the cyclisation of N-dithio-carbalkoxyglycines gave an oil which exhibited only weak colour reactions with an alcoholic-triethylamine solution of a variety of 2-acetanilidovinyl derivatives of cyclic quaternary

^{*} This compound has already been made by Kendall and Duffin (B.P. 633,736).

ammonium salts. On the other hand treatment of (VIII; R = Me) with phosphorus tribromide in dioxan-ether gave a product, the analysis of which indicated the loss of the elements of water, giving the hydrobromide of (IX) or (X). Similar products were obtained from (VIII; $R = Bu^n$ or Ph). All these hydrobromides slowly decomposed in moist air, fuming strongly. They gave strong colour reactions with the above-mentioned quaternary salts, and a number of dimethin*mero*cyanines were isolated (see Experimental). Attempts to isolate the free bases from the hydrobromides by shaking them with aqueous sodium acetate-ether were unsuccessful, the ethereal layer no longer giving any colour reactions. This method of releasing the bases was successfully employed by Cook *et al.* (*loc. cit.*, 1949) in the case of 2-benzylthiothiazol-5-one.

The apparent instability of the bases of these hydrobromides is considered to favour their being thiazolones (X) rather than tetrahydroketoglyoxalines (IX) since the latter would be expected to be stable. Further evidence against (IX) is the non-formation of insoluble silver salts by the dyes derived from the hydrobromides. Thus the thiol group in (IX) (regarded as a mercapto-derivative) should be capable of salt formation as has been found for similar dyes derived from 2-mercaptothiazol-5-one. From an electronic standpoint it would, moreover, be expected that thiazolone formation would be preferred. Thus it may be assumed that the first step in the cyclisation is the formation of the acid bromide (XI) in which one reactive centre is the electrophilic carbonyl-carbon atom of the COBr residue. For formation of (IX) it is then required that this centre should attack the amido-nitrogen atom which is, however, also electrophilic by virtue of the amide-type resonance in the system. For thiazolone formation the nucleophilic sulphur atom is the second active centre and consequently the formation of a covalent C-S bond appears the more likely. Similar electronic conditions also apply in the condensation of thiourea with α -halogeno-ketones or 2-bromoethylamine, the resultant products in such cases being derivatives of 2-aminothiazole or 2-aminothiazoline and not of a di- or tetra-hydroglyoxaline. In view of the sensitivity of 5-imino-2-thiothiazolidines to alkali (Cook et al., J., 1948, 201) (causing isomerisation to 2: 4-dithiohydantoin) there appears the possibility that the compound obtained by Wheeler et al. (loc. cit.) was, in fact, a thiazolidone, the hydrolysing treatment with alkali causing a similar isomerisation to the 2-thiohydantoin derivatives.

The hydrobromide from (VIII; $R = Pr^n$) condensed readily with diazoaminobenzene to give what is considered to be 2-*n*-butyramido-5-hydroxy-4-phenylazothiazole (IV; $R = NH \cdot COPr^n$).

2-Alkoxythiacol-5-ones (XV).—For the preparation of these derivatives, N-thiocarbalkoxyglycines (XIV) were required. They cannot be obtained in the same way as the N-dithioanalogues, but are readily obtained according to the following reaction scheme:



The required alcohol (ROH) was converted into the potassium alkoxydithioformate (alkylxanthate) (XII), esterified with ethyl bromide to give the ethyl alkoxydithioformate (XIII) (omission of this step lowers considerably the yield in the next step), and the latter refluxed with aqueous-alcoholic potassium aminoacetate to give the required N-thioncarbalkoxyglycine (XIV) with the elimination of ethanethiol. Twelve alkyl, and also the 3-phenylpropyl, esters were thus prepared; they are all stable, alkali-soluble crystalline solids, the longer-chain esters being surface-active. They are readily cyclised by acetic anhydride at 100—130° to a series of 2-alkoxythiazol-5-ones. The crude oils were used in all cases for the preparation of a series of dimethin*mero*cyanines. 2-Ethoxythiazol-5-one was distilled without decomposition but the product, even after fractionation, had the odour of acetic anhydride and gave low values for nitrogen and sulphur.

By cyclising in the presence of ethyl orthoformate 2-alkoxy-4-ethoxymethylenethiazol-5ones (XVI) were obtained and the crude products used in dye formation. They are all crystalline products but those with short alkyl chains rapidly decompose and liquefy. The stability increases with increasing chain length and several of these compounds were characterised. The attempted cyclisation of (XIV) by phosphorus tribromide or phosphorus trichloride in dioxan-benzene did not give the required 2-alkoxythiazol-5-one salt. The crystalline product which resulted was well defined and its nature and properties will be reported separately.

2-tert.-Aminothiazol-5-ones (XX).—In view of the failure to bring 2-ethylthiothiazol-5-one into reaction with secondary amines in order to produce 2-tert.-aminothiazol-5-ones their synthesis was attempted by the cyclodehydration of NN-disubstituted N'-carboxymethyl-thioureas (XIX). In theory these may be obtained by two routes. It is well known (Delépine, Compt. rend., 1902, **134**, 715) that secondary amines (NHRR') condense with carbon disulphide to give salts of NN-disubstituted dithiocarbamates (XVII) which are readily esterified by alkyl halides to give (XVIII). Treatment of these esters with glycine would then be expected to give the amide (XIX) but it appears that amide formation does not readily occur in these cases, although it occurs much more readily with N-monosubstituted dithiocarbamates (XVIII; R = H).



It was found then that the required substances were smoothly formed by the condensation of the secondary amine in aqueous or aqueous-alcoholic solution with the potassium salt of N-dithio-carbethoxyglycine (V; R'' = Et). In this way N-(NN-diethylthiocarbamyl)glycine (XIX;

$$(XXI.) \qquad N \cdot CS \cdot NH \cdot CH_2 \cdot CO_2H \qquad O \qquad N \cdot CS \cdot NH \cdot CH_2 \cdot CO_2H \quad (XXII.)$$

R = R' = Et), and the analogues (XIX; R = Ph, R' = Et), (XXI), and (XXII) were obtained. These are cyclised by phosphorus tribromide to oily 2-*tert*.-aminothiazol-5-one hydrobromides which were used directly in dye formation (see Experimental).

EXPERIMENTAL.

Microanalyses are by Drs. Weiler and Strauss, Oxford. M. p.s are uncorrected.

4-1'-Ethoxyethylidene-2-phenyloxazol-5-one (III; R' = Me, X = O, Y = Ph).—Hippuric acid (72 g., 0.4 mol.), ethyl orthoacetate (65 g., 0.4 mol.), and acetic anhydride (80 c.c.) were heated together in an oil-bath at 140—150° for 1 hour and the solvents removed under reduced pressure. The residual red oil crystallised on chilling. It was dissolved in warm ethanol (50 c.c.) and then chilled and the crystals were collected and washed with light petroleum (b. p. 60—80°). The ketone (21 g.) formed pink blades, m. p. 110°, from ethanol (Found : N, 5.95. $C_{13}H_{13}O_3N$ requires N, 6.05%).

N-Dithiocarbalkoxyglycines.—N-Dithiocarbonitrobenzyloxyglycines (V; $R'' = CH_2 \cdot C_6H_4 \cdot NO_2$). Potassium hydroxide (11·2 g., 0·2 mol.) was dissolved in water (25 c.c.), the solution was chilled, and glycine (7·5 g., 0·1 mol.) and carbon disulphide (7·6 g., 0·1 mol.) were added. The mixture was shaken until homogeneous (3—4 hours) and *m*-nitrobenzyl bromide (21·6 g., 0·1 mol.) dissolved in ethanol (25 c.c.) was added. After further shaking for 4 hours the yellow solid (not investigated) was filtered off and the filtrate acidified to give an oil which soon crystallised. It was redissolved in 2N-sodium carbonate, and the solution filtered and acidified. The precipitated m-*nitrobenzyl* ester was collected, washed and air-dried. It (14 g.) formed pale yellow needles, m. p. 144°, from benzene (Found : S, 22·5. $C_{10}H_{10}O_4N_2S_1$ requires S, 22·4%). The p-*nitro*-isomer was similarly obtained from p-nitrobenzyl chloride (17·15 g., 0·1 mol.). The crude product was crystallised by dissolving it in acetone, adding benzene (2 vols.), and slowly distilling off the acetone. It (20·5 g., 72%) formed pale yellow needles, m. p. 146°, from toluene (Found : S, 22·75%).

N-Dithiocarbo-n-heptyloxyglycine (V; $R'' = n-C_7H_{16}$). A solution of potassium hydroxide (56 g., 1.0 mol.) and glycine (37.5 g., 0.5 mol.) in water (100 c.c.) was shaken for 3 hours with carbon disulphide (38 g.). *n*-Heptyl bromide (90 g., 0.5 mol.) and industrial alcohol (100 c.c.) were added and the two-phase mixture was refluxed on the steam-bath for 30 minutes. Before the whole finally became homogeneous a third liquid phase separated. The solution was chilled, water (200 c.c.) was added, the resultant cloudiness removed by ether-extraction, and the aqueous layer acidified. The precipitated oil was extracted with ether, the ethereal solution dried, and the solvent distilled off. The residual oil was disolved in light petroleum (b. p. 80—100°) (300 c.c.) and the solution chilled. There were obtained 57 g. (46%) of glittering, waxy flakes, m. p. 99°, after a second recrystallisation from light petroleum (Found : S, 25.65. $C_{10}H_{19}O_2NS_2$ requires S, 25.7%).

By proceeding similarly the esters recorded in Table I were obtained.

2-p-Nitrobenzylthiothiazol-5-one Hydrobromide.—This was obtained in quantitative yield by dissolving (V; $\mathbf{R}^{\prime\prime} = \cdot \mathbf{CH}_2 \cdot \mathbf{C}_6 \mathbf{H}_4 \cdot \mathbf{NO}_2$) (20 g.) in dioxan (80 c.c.) and adding ether (100 c.c.) and then phosphorus tribromide (12 c.c.). The required salt separated as an oil which rapidly crystallised. It was finely ground under ether and air-dried (Found : Br, 22·3. $\mathbf{C}_{10}\mathbf{H}_8\mathbf{O}_3\mathbf{N}_2\mathbf{S}_2$, HBr requires Br, 22·9%). The m-nitro-isomer was similarly obtained (Found : Br, 22·4%). They both had indefinite m. p.s.

2-Alkylthio-4-1'-ethoxyethylidenethiazol-5-ones and 2-alkylthio-4-1'-ethoxypropylidenethiazol-5ones were obtained by heating the N-dithiocarbalkoxyglycine (15—20 g.), acetic anhydride (75 c.c.), and ethyl orthoacetate or ethyl orthopropionate (30 c.c.) for 30 minutes in an oil-bath at 130°. Removal of the solvents under reduced pressure gave crude oils which were used directly in dye condensations.

2-n-Dodecylthio-5-hydroxy-4-phenylazothiazole (IV; $R = \cdot S \cdot C_{12}H_{25}$).—To a solution of N-dithiocarbon-dodecyloxyglycine (3·2 g.) in dioxan (10 c.c.) and ether (30 c.c.) was added phosphorus tribromide (2 c.c.), and the whole was refluxed for 3 minutes. The solution was chilled and the required 2-ndodecylthiothiazol-5-one hydrobromide precipitated as an oil by the addition of light petroleum (b. p. $40-60^{\circ}$; 100 c.c.). The oil was washed with ether by decantation and dissolved in ethanol (20 c.c.), and diazoaminobenzene (2 g.) was added. The solution was heated on the steam-bath (*a*. 1 minute) until effervescence commenced, the flask removed from the bath, and, after being chilled overnight, the *dye* was collected. It (1·4 g.) formed glossy golden plates, m. p. 83°, from light petroleum (b. p. 60-80°) (Found : N, 10·15. $C_{21}H_{31}ON_3S_2$ requires N, 10·4%).

TABLE I.

R" in (V).	Appearance.	М. р.	Solvent.*	Formula.	Found, S, %.	Reqd., S, %.
n-Hexyl	Soft, glistening plates	95°	Ligroin	$C_9H_{17}O_2NS_2$	27.3	$27 \cdot 2$
n-Octyl	Glossy, waxy plates	100	Light petroleum	$\mathrm{C_{11}H_{21}O_2NS_2}$	23.95	24.35
n- <i>Decyl</i>	Glittering plates	100	Benzene-light petroleum	$\mathrm{C_{13}H_{25}O_2NS_2}$	22.5	$22 \cdot 1$
n-Dodecyl	Glossy plates	111	Ligroin	C15H200NS2	20.3	20.05
3-Phenylpropyl	Waxy plates	122	Benzene-light petroleum	$C_{12}H_{15}O_2NS_2$	24.5	24.8

* Ligroin had b. p. 100-120°. Light petroleum had b. p. 60-80°.

Ethyl n-Butyramidodithioformate (VII; $R = Pr^n$).—Ethyl dithiocarbamate (ethyl aminodithioformate) (Delépine, Compt. rend., 1902, **135**, 975) (25 g.) and butyric anhydride (36 c.c.) were heated at 120° in an oil-bath for 4 hours and the solvents removed under reduced pressure. The residue crystallised on chilling. It was dissolved in methanol at 25° and water run in until slightly turbid. After being seeded and cooled to 5° the ester (29.5 g., 75%) separated as long, yellow needles. A sample formed lemonyellow needles, m. p. 45°, from aqueous methanol (Found : S, 16.55. $C_7H_{13}ONS_2$ requires S, 16.75%).

N-N'-Butyrylthiocarbamylglycine (VIII; $R = Pr^n$).—The above ester (25 g.) was added to a solution of potassium hydroxide (7.5 g.) and glycine (10 g.) in water (30 c.c.) and alcohol (30 c.c.), and the whole refuxed on the steam-bath for 8 hours. The clear solution was concentrated to half volume, water (50 c.c.) added, and the mixture acidified with concentrated hydrochloric acid. The bulky white precipitate was collected, washed with water, and obtained as cream-coloured needles (16 g., 60%), m. p. 180°, from aqueous ethanol (Found : S, 15.95. $C_7H_{12}O_3N_2S$ requires S, 15.7%).

2-Acylamidothiazol-5-one Hydrobromides.—The NN'-acylthiocarbamylglycine (10 g.) was dissolved in dioxan (50—75 c.c.) and ether (100—150 c.c.), and phosphorus tribromide (518 c.c.) added. The required hydrobromide separated rapidly in ca. 95% yield. After 3 hours the solid was collected, washed with ether, and vacuum-dried. They all formed colourless powders which fumed in moist air and gradually decomposed. 2-Acetamidothiazol-5-one hydrobromide has m. p. 195° (decomp.) (Found : N, 11.9; Br, 32.9; S, 13.7. $C_5H_4O_2N_2S$,HBr requires N, 11.7; Br, 33.45; S, 13.4%). 2-Butyramidothiazol-5-one hydrobromide has m. p. 220° (decomp.) (Found : Br, 29.5. $C_7H_{10}O_2N_2$,HBr requires Br, 30.0%). 2-Benzamidothiazol-5-one hydrobromide has m. p. 190° (decomp.) (Found : Br, 24.7. $C_{10}H_8O_2N_2S$,HBr requires Br, 26.6%).

2-Butyramido-5-hydroxy-4-phenylazothiazole (IV; $R = NHPr^{n}$).—2-Butyramidothiazol-5-one hydrobromide (1 g.), diazoaminobenzene (1.35 g.), and ethanol (10 c.c.) were warmed together on the steambath until effervescence occurred (1—2 minutes). When the mixture was then set aside the *dye* crystallised and formed golden-orange crystals (0.3 g.), m. p. 251°, from ethanol (Found : N, 19.4. $C_{12}H_{14}O_2N_4S$ requires N, 19.8%).

N-Thiocarbalkoxyglycines (XIV).—The potassium alkoxydithioformates (xanthates) were obtained by the method of de la Provostarge and Desains (Compt. rend., 1942, **215**, 593) by dissolving the required alcohol (1 mol.) in carbon disulphide (100—250 c.c.), adding finely powdered potassium hydroxide (1 mol.), and shaking the mixture for 1—2 hours. After chilling, the crude filtered salts were washed with ether and air-dried. They were converted into the esters (ethyl alkoxydithioformates) by addition of ethyl bromide (1 mol.) to a suspension of them (1 mol.) in ethanol and warming if necessary to start the reaction. The thick meal of potassium bromide was dissolved in water, the oily ester extracted with ether and dried, and the ether removed (cf. Salomon, J. pr. Chem., 1872, **6**, 445). The crude oils were used directly in the next step.

The following illustrates the general procedure for the final step. N-Thioncarbethoxyglycine (XIV; R = Et). Ethyl ethoxydithioformate (75 g., 0.5 mol.) in alcohol (50 c.c.) was added to a solution of

potassium hydroxide (28 g., 0.5 mol.) and glycine (37.5 g., 0.5 mol.) in water (50 c.c.), and the whole refluxed on the steam-bath for 18 hours, the ethanethiol being allowed to distil off. Water (50 c.c.) was added, and the solution clarified by ether-extraction, concentrated to half volume, and acidified with cold concentrated hydrochloric acid (50 c.c.). The required *ester* (55.5 g., 68%) separated and formed glassy needles, m. p. 101°, from water (Found : C, 37.0; H, 5.0; N, 8.2; S, 19.65. $C_5H_9O_3NS$ requires C, 36.8; H, 5.5; N, 8.6; S, 19.9%).

The esters recorded in Table II were obtained in a similar manner.

2-Alkoxythiazol-5-ones.—N-Thioncarbalkoxyglycine (10 g.) and acetic anhydride (50 c.c.) were heated for 30 minutes in an oil-bath at 130°, and the solvents removed under reduced pressure. The residual oils were used directly in dye formation. *E.g.*, *N*-thioncarbethoxyglycine (10 g.) and acetic anhydride (50 c.c.) gave an oil which on distillation gave a colourless oil, b. p. 57°/4 mm. (62°/5 mm.), which became light brown on storage (Found : N, 7.65; S, 18.6. $C_5H_7O_2NS$ requires N, 9.65; S, 22·1%). It gave the same dyes as the crude reaction mixture and appears to be the required 2-ethoxythiazol-5-one contaminated with acetic anhydride.

			TABLE II.			
R in (XIV).	Appearance.	М. р.	Solvent.*	Formula.	Found, S, %.	Reqd., S, %.
isoPropyl	Long, flat needles	129°	Water	$C_6H_{11}O_3NS$	18.1	18.1
n-Butyl	Glossy needles	70-71		C,H,O,NS	16.95	16.75
n-Amyl	Needles	49 - 50	Light	C ₈ H ₁₅ O ₃ NS	15.75	15.6
			petroleum	0 10 0		
tertAmyl	Fine threads	52 - 54	~ ,,	,,	15.75	
n-Hexyl	Small needles	62	,,	C ₉ H ₁₇ O ₃ NS	6.25 (N)	6·4 (N)
n-Heptyl	,,	66	,,	C ₁₀ H ₁₉ O ₃ NS	6·15 (N)	6·0 (N)
n-Octyl	Fine needles	72	,,	$C_{11}H_{21}O_3NS$	5·8 (Ň) ́	5·65 (Ń)
n-Decyl	Glistening plates	63	,,	$C_{13}H_{25}O_3NS$	11.55	11·65 `´
n-Dodecyl	Fine needles	67	,,	C ₁₅ H ₂₉ O ₃ NS	4·85 (N)	4·6 (N)
n-Tetradecyl	,,	79	,,	$C_{17}H_{33}O_{3}NS$	9·5 `´	9.65
n-Hexadecyl	,,	77	,,	$C_{19}H_{37}O_{3}NS$	4·05 (N)	3.9(N)
3-Phenylpropyl	Asbestos-like threads	88	Benzene-light petroleum	$C_{12}H_{15}O_3NS$	12.5	12.65

* Light petroleum had b. p. 60-80°.

4-Ethoxymethylene-2-alkoxythiazol-5-ones.—The N-thiocarbalkoxyglycine (0.01 mol.), acetic anhydride (10 c.c.), and ethyl orthoformate (3.5 c.c.) were heated in an oil-bath at 135— 140° for 30 minutes, the alcohol being allowed to distil off. Removal of the solvents under reduced pressure left an oil which usually crystallised on chilling. The crude products were used directly in dye formation. The following were characterised :

2-n-Decyloxy-4-ethoxymethylenethiazol-5-one, pale yellow needles, m. p. 50°, from ethanol (Found : N, 4·4. $C_{16}H_{27}O_3NS$ requires N, 4·45%); 2-n-dodecyloxy-4-ethoxymethylenethiazol-5-one, pale yellow needles, m. p. 58°, from ethanol (Found : N, 4·1. $C_{18}H_{31}O_3NS$ requires N, 4·1%); 4-ethoxymethylene-2-n-tetradecyloxythiazol-5-one, soft, pale yellow needles, m. p. 64°, from ethanol (Found : C, 65·1; H, 9·25; N, 3·5; S, 8·65. $C_{20}H_{35}O_3NS$ requires C, 65·0; H, 9·5; N, 3·8; S, 8·7%).

N-(NN-diethylthiocarbamyl)glycine (XIX; R = R' = Et).—Potassium hydroxide (6·24 g.) was dissolved in water (20 c.c.), N-dithiocarbethoxyglycine (20 g.) and diethylamine (11·8 c.c.) were added, and the whole was refluxed for 15 hours. After cooling, the solution was extracted with ether, and the aqueous layer cautiously acidified with concentrated hydrochloric acid. After chilling and scratching, crystallisation set in. The solid (11·0 g., 52%) after air-drying formed small, buff-coloured crystals, m. p. 105°, from benzene (Found : N, 14·4; S, 16·8. $C_7H_{14}O_2N_2S$ requires N, 14·7; S, 16·85%).

N-(N-Methyl-N-phenylthiocarbamyl)glycine (XIX; R = Ph, R' = Et).—Potassium hydroxide (3·12 g.) and N-dithiocarbethoxyglycine (10 g.) were dissolved in water (16 c.c.) and ethanol (32 c.c.), and after the addition of N-ethylaniline (6·8 g.) the whole was refluxed 24 hours on the steam-bath. Water (50 c.c.) was added and the mixture extracted with ether. The aqueous layer was acidified with concentrated hydrochloric acid and the *amide* collected after chilling. It (3·05 g., 21%) formed a buff-coloured crystalline powder, m. p. 138°, from water (Found : N, 11·8. $C_{11}H_{14}O_2N_2S$ requires N, 12·0%).

1-N'-Carboxymethylthiocarbamylpiperidine (XXI).—Potassium hydroxide (18.72 g.) and N-dithiocarbethoxyglycine (60 g.) were dissolved in water (250 c.c.), to this solution was added piperidine (28.5 g.), and the mixture refluxed for 24 hours. Acidification of the chilled solution gave a bulky precipitate of the glycine derivative as a hydrate (18.5 g., 22.6%) which formed colourless needles, m. p. 171°, from water (Found : N, 12.9; S, 14.3. $C_8H_{14}O_2N_2S,H_2O$ requires N, 12.6; S, 14.55%).

4-N'-Carboxymethylthiocarbamylmorpholine (XXII).—Obtained similarly from morpholine (29·2 g.), this formed a hydrate, colourless needles, m. p. 171° (49·5% yield), from water (Found : N, 12·3; S, 14·2. $C_7H_{12}O_3N_2S,H_2O$ requires N, 12·6; S, 14·4%).

The 2-tert.-aminothiazol-5-one hydrobromides were obtained as oils on addition of phosphorus tribromide (1 c.c.) to the above glycines (2 g.) dissolved in dioxan (15 c.c.) and anhydrous ether (45 c.c.). The oils were washed with ether and used directly in the dye condensations.

Dimethinmerocyanines.[¶]—The dyes recorded in Tables III and IV were prepared by treating, for 5—10 minutes on the steam-bath, an alcoholic solution of triethylamine (1 mol.) and (a) the 2-acetanilido b_{a} in the steam-bath, an arconom solution of the try minine (1 mol.) and (a) the 2-acetanlido-vinyl derivative of the required cyclic ammonium salt (1 mol.) with the keto-methylene heterocyclic compound (II) (1 mol.) or (b) the reactive methyl derivative of the same ammonium salt (1 mol.) with the 4-ethoxymethylene (or 4-1'-ethoxyethylidene) derivative (III) (1 mol.). When the hydrobromides of the latter were employed two mols. of triethylamine were added.

TABLE III.

	[Heterocyclic nucleus][4-(2-phenyl-	-					
Ref.	oxazol-5-one)] dimethinmero-	Appear-				Found,	Reqd.,
no.	cyanine.	ance.	М. р.	S.‡	Formula.	N, %.	N, %.
1	[2-(1-Ethylquinoline)] *	Green plates	227°	BL	$\mathrm{C_{22}H_{18}O_2N}$	7.85	8 ·2
2	[2-(3-Ethylbenzoxazole)] *	Red needles	225	BL	$C_{20}H_{16}O_{3}N_{2}$	8.35	8.45
3	[2-(3-Ethylbenzothiazole)] *	,,	209 +	BL	C ₂₀ H ₁₆ O ₂ N ₂ S	7.75	8.05
4	[2-(3-Ethylbenzoselenazole)]	Magenta needles	222	BL	$C_{20}H_{16}O_2N_2Se$	6·8	$7 \cdot 1$
5	[2-(1:3:3-Trimethylindolenine)]	Red needles	198	BL	$C_{22}H_{20}O_2N_2$	7.5	8.15
6	[4-(1-Ethylquinoline)]	Green	212	BL	$\rm C_{22}H_{18}O_2N_2$	7.85	$8 \cdot 2$
7	[2-(3-Methylthiazoline)] *	Brick-red	212	м	$C_{15}H_{14}O_2N_2S$	9.9	9.8
8	[2-(3-Ethyl-4-methylthiazole)]	Steel-grey	209	BL	$C_{17}H_{16}O_2N_2S$	9.05	9.0
9	[2-(3-Methylnaphtho-1': 2'-4: 5-oxazole)]	Red	288	PW	$C_{23}H_{16}O_{3}N_{2}$	$7 \cdot 2$	7.6
10	[2-(3-Ethylnaphtho-2': 1'-4: 5-	,,	250	В	$C_{24}H_{18}O_3N_2$	7.35	7.35
11	[2-(3-Ethylnaphtho-1': 2'-4:5-	Purple	206	В	$C_{24}H_{18}O_2N_2S$	7 ·0	7.05
12	[2-(3-Ethylnaphtho-2': 1'-4:5-	",	260	\mathbf{PW}	,,	7.0	7.05
13	[2-(1-Ethyl-7 : 8-benzoquinoline)]	Green	139	BL	$C_{26}H_{20}O_2N, C_6H_0$	6 ∙0	5.95
14	[2-(3-Ethyl-4-p-methoxyphenyl- thiazole)]	Red needles	180	В	$C_{23}H_{20}O_{3}N_{2}S$	6.8	6.95

* Obtained by Keyes and Brooker (*loc. cit.*). † Cook *et al.*, *loc. cit.*, give m. p. 210°. ‡ Solvent : B = benzene; C = chloroform; E = ethanol; L = light petroleum (b. p. 40-60°); M = methanol; P = pyridine; W = water.

TABLE IV.

		1110000 1					
Ref. no.	[a-(Heterocyclic nucleus][eta-4(2-phenyloxazol-5-one)]-eta-methyl-dimethinmerocyanine.	Appear- ance.	М. р.	S.*	Formula.	Found, N, %.	Reqd., N, %.
15	[a-2-(1-Ethylquinoline)]	Flat, green needles	198°	BL	$C_{23}H_{20}O_2N_2$	7.55	7.85
16	[a-2-(3-Methylthiazoline)]	Brick-red crystals	210	BL	$C_{16}H_{16}O_2N_2S$	9.1	9.35
17	[a-2-(3-Ethylbenzoxazole)]	Orange needles	209	BL	$C_{21}H_{18}O_{3}N_{2}$	7.7	8.1
18	[a-2-(3-Ethylbenzothiazole)]	Red needles	237	в	$C_{21}H_{18}O_2N_2S$	$7 \cdot 9$	7.75
19	[a-2-(3-Ethylbenzoselenazole)]	Violet	228	в	$C_{21}H_{18}O_2N_2Se$	6.95	6.85
20	$[a-2-(3-Ethyl-4-methylthiazole)] \dots$	Purple needles	232	В	$C_{18}H_{18}O_2N_2S$	8.9	8.6

* See footnote, table III.

Dimethinmerocyanines (cf. Table V) containing the 2-alkylthiothiazol-5-one nuclei were obtained similarly. *mero*Cyanines (ii) Table V) containing the standard procedure of treating the thiazolone (II) with the 2-methylthio-derivative of the cyclic ammonium salt in alcoholic triethylamine. *Tetramethin*-mero*cyanines* (Table V) were obtained similarly by using the 4-acetanilidobuta-1: 3-dienyl derivatives of the cyclic ammonium salt.

[a-2-(3-Ethylbenzoxazole)][β-4-(2-n-decylthiothiazol-5-one)]-β-methyldimethinmerocyanine.—N-Dithiocarbo-n-decyloxyglycine (1.45 g.), acetic anhydride (15 c.c.), and ethyl orthoacetate (6 c.c.) were heated

[¶] In the naming of the dyes the ketonic nucleus is placed in the second bracket irrespective of the alphabetical order of the two nuclei. The nucleus in the first bracket, which is actually a dihydro-derivative of the heterocyclic nucleus, is simplified in all cases. Thus 3-ethylbenzothiazoline is called 3-ethylbenzothiazole, 3-ethylthiazolidine is called 3-ethylthiazoline, etc., as the termination *mero*cyanine indicates the structure. Hamer and Winton (*J.*, 1949, 1126) use both forms. The first dye in Table III is thus [2-(1-ethylquinoline)][4-(2-phenyloxazol-5-one)]dimethin*mero*cyanine.

	219	92		A	lul	ber	νt,	ŀ	(n	iot	t,	а	nc	l	W	'il	li	an	ıs	÷	1	Di	m	etk	iir	ım	let	:00	cyd	an	in	les							
	Reqd., N, %.		8.75 8.75	8-65 8-45	8.05	1.7	10.65 +	0.6	6.95 1 0 1	00-7 2-8	50 00 00 00	8.15	7.1	6-85 6.1	1.0 9	8.05	6.95	6.95	8.1 8.95	10.7	9.6	10.7	0.20 9.6	16.5 †	15.9 +	6-05 95.05 +	25.0 +	6.75	22-2 + 90.6 +	5.85	6.2	13.55	12.01	r		7-55		25-65 †	
	Found, N, %.	- c	1.8 2.9	8.65 8.45	100 100 100	7-05 9-6	10.75 +	9.1	6.75	1.1	8·1	8.1	7.05	6.4 7.05	0.9. 2.9	7.95	6.55	6.75	8.7	10.45	9.2	10-7	9.4 0.4	$16.6 \pm$	15.4 †	6.1 96.2 +	25.25 +	6.75	22.2 +	6-05	5.95	13.55	1 eu-ei	u r r	61.1	7-45		$25.8 \pm$	
	Formula.		$C_{14}H_{12}O_{N_2}S_3$ $C_{14}H_{12}O_{N_2}S_3$	C16H14ON252 C16H14ON252	C16H16ON2S3	C16H16ONSSSE	C.,H.,ON,S.	C ₁₃ H ₁₆ ON ₂ S ₃	C10H100NS3	C ₂₀ H ₁₈ CN ₂ S ₃ C ₁₀ H ₁₀ ONS	V181118/112/22	$C_{18}H_{20}ON_2S_2$	$C_{20}H_{16}O_2N_2S_2$	C ₂₁ H ₁₈ ON ₂ S ₃	C ₂₁ H ₁₈ ON ₂ S ₂ Se	C ₁₆ H ₁₆ ON ₂ S ₃	C ₂₃ H ₂₀ ON ₂ S ₂	C ₂₃ H ₂₀ ON ₂ S ₂		C1811601303	$C_{21}H_{17}O_{4}N_{3}S_{2}$	C16H15O3N3S3		$C_{20}H_{34}O_{3}N_{3}S_{3}$	C ₂₁ H ₂₆ O ₂ N ₂ S ₂	C ₂₁ H ₂₆ ON ₂ S ₂ Se	C171126 CN 303 C1RH 30 N S3	C22H28O2N2S2	C ₂₂ H ₂₈ ON ₂ S ₃	C.H.ON.S.Se	C24H3202N2S2	C ₂₆ H ₃₆ O ₂ N ₂ S ₂	C23H22U2H252		C20H16CIN202	C ₁₈ H ₁₄ ON ₂ S ₃		$C_{18}H_{18}ON_2S_3$	te to Table III.
۶,	s.‡	¢	٩œ،	ਸ	'nА	ц	न त्म	ы	фр	д НЦ	ц	$\overline{\mathrm{BE}}$	щ	피며	а ВЕ	BE	Щ	ы	리伍	PE	PE	PE	되 전 전 전	ц ц	ĽЦ	피ഥ	비머	ы	ы МЫ	BM	ы	Мр	चे	זם	DL	BL	ļ	म म	see tootno
C SI	M. p.	0000	236	156	173	184 139	121	154	182	1/2	143	137	193	179 165	215	171	207	162	179	202	201	187	199	148	133	112 81	86	130	81 156	88	106	103	0/T	144	144	193	1	179	,, ++
C:[CH-CH];	v. R Appearance.	lanine.	Erown crystals	Black powder Flat. steel blue needles	Blue-grey prisms	,, Maroon needles	Yellow plates	Green prisms	Maroon needles	,, Green needles	Blue plates	Red needles	Orange needles	Ked needles	Purple needles	Red-green prisms	Green needles	Blue needles	Furple needles Green crystals	Bronze plates	Purple needles	Steel-blue plates	Green needles Red crystals	Red-brown platelets	Rust-red crystals	Glassy, violet needles Durnle needles	Orange crystals	Glossy orange crystals	Red needles	Violet needles	Orange powder	Rust powder		I = 0. D_{0-1} more second 1	Dark-green crystals	Yellow needles	ocyanine $(n = 2)$.	Blue crystals	Sulphur (not nitrogen).
E	IABLE R'.	ne)]dimethinmeroc3	6 *	Ĕť			: :	: :		•	. :		CH_2Ph			: :	2		:	CH".C.H.NOb		CH2.℃6H4.NO2-m		C ₈ H ₁₃ - <i>n</i>	$C_{7}H_{15}-n$	С Ц'_"	~~1118~		:		$C_{10}H_{21}'n$	$C_{13}H_{35}-n$	[сп ₂] ₃ ти	<i>ne)</i>]mero <i>cyanne</i> (I CII Dh	CH2FII	:	me)]tetramethinmer	Et	÷.
	Heterocyclic group.	[Heterocyclic group][4-(2-alkylthiothiazol-5-0	[2-(3-Ethylbenzoxazote)] [2-(3-Ethylbenzothiazole)]	[2-(1-Éthylquinoline)] [2-(3-Ethvlhenzoxazole)]	[2-(3-Ethylbenzothiazole)] *	[2-(3-Ethylbenzoselenazole)]	[2-(3-Ethvlthiazoline)]	[2-(3-Ethyl-4-methylthiazole)]	[2-(4-p-Methoxyphenyl-3-ethylthiazole)]	[2-(3-Ethylnaphtho-1': 2'-4: 5-thiazole)] [9_/1_E+hyloninoline\]	[2-(1-Ethylaninoline)]	[2-(1:3:3-Trimethylindolenine)]	[2-(3-Ethylbenzoxazole)]	[2-(3-Ethylbenzothiazole)] *	[2-(3-EurylDenzOselenazOle)] [9_(3-Fthwinamhtho_1'· 9'-4 · 5-thiazole)]	[2-(3-Methylthiazoline)]	[2-(1-Ethylquinoline)] [[4-(1-Ethylquinoline)]	[2-(1-Ethylpyridine)]	[±-(1-Methylpyliume)]	[2-(3-Ethylbenzoxazole)]	[2-(3-Methylthiazoline)]	[2-(3-Ethylbenzothiazole)]	[2-(3-Ethylbenzoxazole)]		[2-(3-Ethylbenzoselenazole)]	[2-(o-metu) umazoune)][2-(3-Ethvlthiazoline)]	[2-(3-Ethylbenzoxazole)]	[2-(3-Ethylbenzothiazole)]	[2-(3-Ethylbenzoselenazole)]	[2-(3-Ethylbenzoxazole)]			[Helevocyclub group][4-(2-alkyllniotniazol-0-6	[z-(1-Metuyiquinoune)]	[2-(3-Methylbenzothiazole)]	[Heterocyclic group][4-(2-alkylthiothiazol-5-0	[2-(3-Ethylbenzothiazole)]	* See also Cook et al. (loc. cit.
	Ref. no.	Ş	222	23 7 7 23	122	26 26	287	29	30	31 2	332	34	35	36	20	30	40	41	47 77 0	£4	45	46	47	40 49	50	51	23	54	50 10 10	270	580	59	00	1.0	ΓQ	62		63	

11: • **,** .

		Substituted 11201-5-01105.	
	Redd., N, %. 12.4 12.4 12.15 11.75 11.75 11.75 11.75 12.2	N, %, %, %, %, %, %, %, %, %, %, %, %, %,	
	Found, N, %. 12:15 12:15 12:35 12:35 12:4 12:4 12:1 12:4 11:9	$\substack{F_{0und,}\\ N, \%, \\ 11.6\\ 9.8\\ 9.8\\ 12.3\\ 9.8\\ 12.8\\ 11.5\\ 11.5\\ 11.5\\ 11.5\\ 11.5\\ 11.6$	
	Formula. C ₁₆ H ₁₇ O ₂ N ₃ S C ₁₆ H ₁₅ O ₂ N ₃ S C ₁₆ H ₁₅ O ₂ N ₃ S C ₁₆ H ₁₅ O ₂ N ₃ S C ₁₁₈ H ₁₀ O ₂ N ₃ S C ₁₁₈ H ₁₁ O ₂ N ₃ S C ₁₁₈ H ₁₁ O ₂ N ₃ S C ₁₁₈ H ₁₅ O ₂ N ₃ S C ₁₆ H ₁₅ O ₂ N ₃ S	Formula. C ₂₀ H ₂₃ ON ₃ S ,, C ₁₈ H ₂₁ ON ₃ S C ₁₈ H ₁₀ O _N S C ₁₈ H ₁₀ C C ₁₈ H ₁₀ O _N S C ₁₈ H ₁₀ O _N S C ₁₈ H	
ĸ	цановасы Хановасы Хановасы Хано	ж 	
N C•NHI	$\begin{array}{c} M.\\ 253,\\ 254,\\ 228,\\ 223,\\ 223,\\ 223,\\ 223,\\ 223,\\ 242,\\ 2$	M. P. C.NR. M. P. C.NR. 158° 163 158° 167 158 158 153 167 158 153 158 153 158° 158° 158° 158° 158° 158° 158° 158°	
Сссн-сн:с	Ř Appearance. Green needles Green powder Red needles Creen-brown needles Orange crystals Green crystals Brick-red needles Orange prisms Brick-red needles Orange needles Orange needles Orange needles Drick-red needles Drick-red needles Drick-red needles Drick-red needles Drick red needles Drange needles	C:CH-CH:C R Appearance. Green needles Green crystals, golden reflex Maroon plates Maroon plates orange-red crystals Maroon needles Orange-red crystals Maroon needles orange prisms Brick-red plates Brick-red plates Brick-red plates Brick-red plates Maroon plates Orange powder	See footnote to Table III.
TABLE VI.	R'. COMe , , , , COPh COPh	[ABLE VIII. NR'R''. NEt ₂ " " NPhEt " " " " " " " " " " " " " "	+÷
	[(Heterocyclic group)]][4-(2-acyl- amidothiazol-5-one)]dimethin- merocyanine. [4-(1-Ethylquinoline)]	 [(Heterocyclic group)][4-(2-tert aminothiazol-5-one)]dimethin- aminothiazol-5-one)]dimethin- aminothiazol-5-one)]dimethin- merocyanine. [2-(1-Ethylquinoline)]	
	A A 99999992222	Re R	

[1951]

	Reqd., N, %. 10:35 23:7 *	9.85 8.55 8.1	8.1 8.35	7.95 7.05	7.35 7.65	9.45 8.5	0-01	8·1 8·15	9-4	9-0	0.6 9.8	8-25 7-05	7.9 6.05	7.35	0-80 6-4	0.0 1.8	
	Found, N, %. 10-3 23.65 *	9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	8.2 8.2	7.6 7.05	7.6 7.6	9.2 8.35 7.7	G16	8·15 8·4	9-35	9-0	9-35 8-5	8.45 6.75	8.1		7-U5 6-35	5.7 7.85	
	Formula. C ₁₁ H ₁₄ O ₂ N ₂ S ₂	C ₁₂ H ₁₆ O ₂ N ₃ S ₂ C ₁₈ H ₂₀ N ₂ S C, H, O,N ₅ S	C ₁₆ H ₁₆ O ₃ N ₂ S	C ₁₆ H ₁₆ O ₂ N ₂ S C ₁₆ H ₁₆ O ₂ N ₂ SSe	C20H18O2N22S2 C20H18O3N5S2	C ₁₃ H ₁₆ O ₂ N ₂ S ₁ C ₁₇ H ₁₈ O ₃ N ₂ S	C12H16O2N252	C1,7H18O2N2S2 C1,2H3,O3N5S	C13H18O2N2S	$C_{14}H_{20}O_2N_2S_2$	C ₁₄ H ₂₀ O ₂ N ₂ S ₂ CH ₂₀ O ₂ N ₂ S ₂	C16H202N2S2 CmH202N2S2	$C_{17}^{22}H_{26}O_{2}N_{2}S_{2}$	C19H302N2S2	$C_{21}H_{34}O_2N_2S_2$ $C_{23}H_{38}O_2N_2S_2$	$C_{16}^{2}H_{12}^{4}O_{2}^{1}N_{2}^{2}S_{2}^{2}C_{16}H_{20}O_{2}N_{2}^{2}S_{2}^{2}$	
	B.t	ыMM	ЖЯ	ലമ	മ്പ	ыы	뇌	ЯĘ	ы	N	ਸ਼ਸ	비비머	ΝĽ	비며	쾨띠	шш	П.
P.O.R.	M. p. 209°	136 170 145	$183 \\ 192$	188 192	202 248	173 184	180	163 150	172	128	136-138 117	111	114-116	115	111	106-107 159	e to Table I
C:CH-CH:CN	Appearance. Brown needles	Orange needles Glossy, red plates Blue needles	Grey-green crystals Golden plates or hard prisms	Violet needles Garnet needles	Purple needles Orange needles	Garnet plates Orange needles	Flat, brown needles, blue reflex	Red needles Orange needles	Flat, brown needles, blue reflex	Flat, brown needles, blue reflex	Flat, red-brown platelets Vellow powder	Orange cryst. Black plates	Flat, yellow needles	Orange powder	Yellow powder Glistening, yellow cryst.	Gold yellow crystals Flat, brown needles	‡ See footnot
BLE VII.	R'. Et	:				ŀř.	:	,, Bu ⁿ	:	n-C ₅ H ₁₁	tert C_5H_{11} <i>n</i> - C_1H_{12}	$n-C_{7}H_{15}$	n - $\ddot{\mathrm{C}}_{8}\mathrm{H}_{17}$	$n \cdot \widetilde{\mathrm{C}}_{10}\mathrm{H}_{21}$	$n-C_{12}H_{25}$ $n-C_{14}H_{20}$	<i>n</i> -C ₁₆ H ₃₃ [CH ₂] ₃ ·Ph	lot nitrogen).
TAI	ef. [(Heterocyclic group)][4-(2-alkoxythiazol- 5-one)]dimethin <i>mero</i> cyanine. 4 [2-(3-Methylthiazoline)]	 [2-(3-Ethylthiazoline)]	8 [4-(1-Ethylquinoline)] 9 [2-(3-Ethylbenzoxazole)]	00 [2-(3-Ethylbenzothiazole)]	32 [2-(3-Ethylnaphtho-1': 2'-4: 5-thiazole)] 33 [2-(3-Ethylnaphtho-1': 2'-4: 5-oxazole)]	34 [2-(3-Ethyl-4-methylthiazole)]	36 [2-(3-Methylthiazoline)]	37 [2-(3-Ethylbenzothiazole)]	89 [2-(3-Methylthiazoline)] ⁻	0 [2-(3-Methylthiazoline)])] ,,)3 33 [4.(1.Fthvlminoline)]	5 [2-(3-Mutry future state)]	00 [2-(0-Eurymenzosetenazote)])8 19	00	* Sulphur (no
	Ϋ́ a Γ			00 00	00 00	00 00	œ	00 00	ŝ	9	00	- 00 0	ິດເ		0 0	201	

together in an oil-bath at 120° for 30 minutes. The solvents were removed under reduced pressure, 2-methylbenzoxazole ethiodide (1.45 g.), ethanol (10 c.c.), and triethylamine (1 c.c.) were added, and the whole was heated for 5 minutes on the steam-bath. The solvent was boiled off and the residue scratched to start crystallisation. The triethylamine hydriodide was removed with a little cold methanol, and the dye was then obtained as soft, orange needles, m. p. 68°, from ethanol (Found : N, 5.95. $C_{25}H_{34}O_{3}N_{2}S_{2}$ requires N, 6.1%). The β -ethyl homologue was similarly obtained by using orthopropionate and formed soft, orange needles, m. p. 96°, from ethanol (Found : N, 6.25. $C_{26}H_{36}O_{2}N_{2}S_{2}$ requires N, 5.95%).

 $[a-2-(3-Ethylnaphtho-1': 2'-4: 5-thiazole)][\beta-4-(2-ethylthiothiazol-5-one)]-\beta-methyldimethinmerocyanine was similarly obtained, formed violet needles, m. p. 209°, from benzene (Found: N, 6.8. C₂₁H₂₀ON₂S₃ requires N, 6.8%), and had an absorption max. at 540 mµ. in methanol.$

Absorption Data.-These are recorded below.

Ref.		Ref.		Ref.		Ref.		Ref.	
no.	λ_{\max} .	no.	$\lambda_{max.}$	no.	λ_{\max} .	no.	λ_{\max} .	no.	λ_{\max} .
1	538 (567i)	24	500 (474i)	47	530 (505i)	70	489 (472i)	93	444
2	492` ´	25	527 (500i)	48	492 (475i)	71	476 (464)	94	568
3	522 (493i)	26	528 (505i)	49	494 (463i)	72	498 (478i)	95	444
4	525 (495i)	27	463 (448)	50	491 (465i)	73	486 (468)	96	500
5	492 (465i)	28	470	51	532 (510i)	74	445	97	445
6	620 (580)	29	528 (502i)	52	478 (463)	75	452	98	444
7	474 (450)	30	530 (505i)	53	479 (464)	76	473	99	445
8	530 (502i)	31	548 (515i)	54	489 (465i)	77	530 (500i)	100	444
9	506	32	546	55	529 (498i)	78	568 (607)	101	445
10	506	33	591 (620i)	56	528 (498i)	79	460	102	560
11	529	34	500 (490)	57	531 (500)	80	499	103	588
12	539 (506i)	35	490 (476i)	58	492 (472i)	81	505	104	480
13	531 (505)	36	530 (508i)	59	490 (476i)	82	541	105	519
14	531 (505)	37	531 (508i)	60	492 (470i)	83	485	106	517
15	566	38	545 (520i)	61	485	84	518	107	560
16	475 (455)	39	472 (462)	62	400	85	467	108	542
17	490 (462i)	40	549	63	640 (596i)	86	446	109	469
18	515 (495i)	41	593 (620i)	64	555 (600)	87	494	110	481
19	518 (492i)	42	510	65	593 (630)	88	469	111	482
20	514 (484i)	43	533	66	488 (462i)	89	450	112	518
21	493	44	478 (460)	67	528 (500i)	90	440	113	480
22	522	45	493 (470i)	68	480 (464)	91	443	114	519
23	551 (582i)	46	478 (465)	69	528 (500i)	92	444	115	479

The authors are indebted to Miss M. E. Cole and Miss E. M. Hamilton for the preparation of some of the intermediates.

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[Received, March 15th, 1951.]